

Paper 1

Questions are applicable for both core and extended candidates

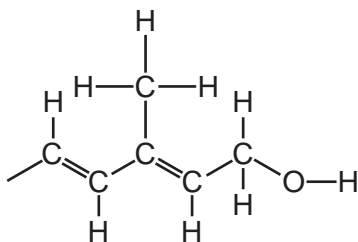
1 Which type of compound is also the name of a homologous series?

- A carbonate
- B carboxylic acid
- C halide
- D hydroxide

2 Which row identifies the homologous series to which the molecular structure belongs?

	molecular structure	homologous series
A	$ \begin{array}{cccccc} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} \\ & & & & & & \\ \text{H} & - \text{C} & - \text{C} & - \text{C} & - \text{C} & - \text{C} & - \text{C} - \text{H} \\ & & & & & & \\ & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} \end{array} $	alkane
B	$ \begin{array}{ccc} & \text{H} & \text{H} \\ & & \\ \text{H} & - \text{C} & - \text{C} - \text{H} \\ & & \\ & \text{H} & \text{H} \end{array} $	alkene
C	$ \begin{array}{ccc} & \text{H} & \text{H} & \text{O} \\ & & & // \\ \text{H} & - \text{C} & - \text{C} & - \text{C} \\ & & & \backslash \\ & \text{H} & \text{H} & \text{O} - \text{H} \end{array} $	alcohol
D	$ \begin{array}{cccc} & \text{H} & \text{H} & \text{H} & \text{H} \\ & & & & \\ \text{H} & - \text{C} & - \text{C} & - \text{C} & - \text{C} - \text{O} - \text{H} \\ & & & & \\ & \text{H} & \text{H} & \text{H} & \text{H} \end{array} $	carboxylic acid

3 Part of the structure of a molecule of vitamin A is shown.



Which statements about this part of the structure are correct?

- 1 It is saturated.
- 2 There are two alkene groups.
- 3 The structure shows a carboxylic acid.

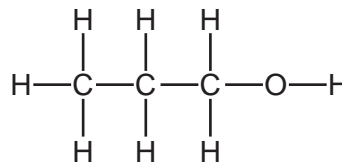
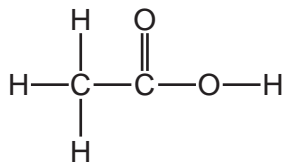
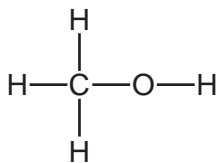
A 1 and 3

B 1 only

C 2 and 3

D 2 only

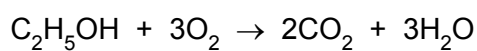
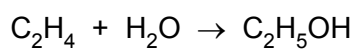
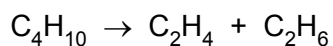
4 The structures of three organic molecules are shown.



Which description of the three molecules is correct?

	they all have the same general formula, $C_nH_{2n+1}OH$	they all belong to the same homologous series
A	no	no
B	no	yes
C	yes	no
D	yes	yes

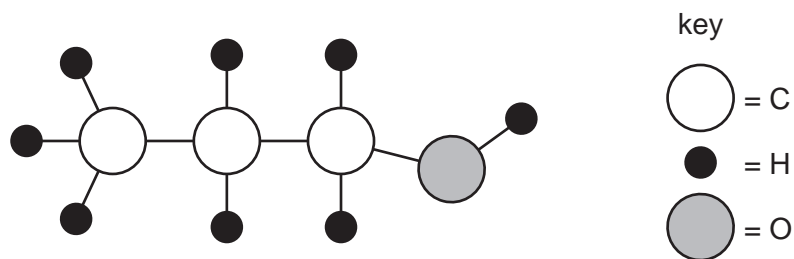
5 Three equations involving organic compounds are shown.



How many different homologous series are shown in these equations?

- A** 1 **B** 2 **C** 3 **D** 4

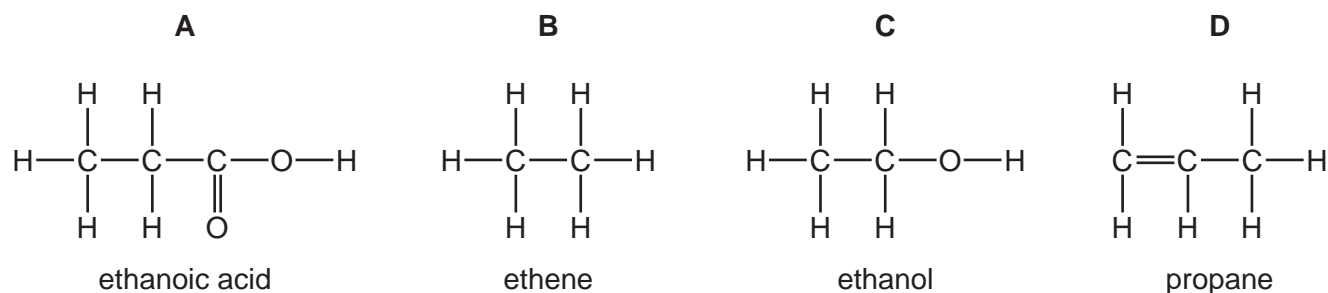
6 The structure of a molecule of a compound is shown.



What is the formula of this compound?

- A** $\text{C}_3\text{H}_7\text{O}$ **B** $\text{C}_3\text{H}_8\text{O}$ **C** $\text{C}_8\text{H}_3\text{O}$ **D** C_8HO_3

7 Which structure is correctly named?

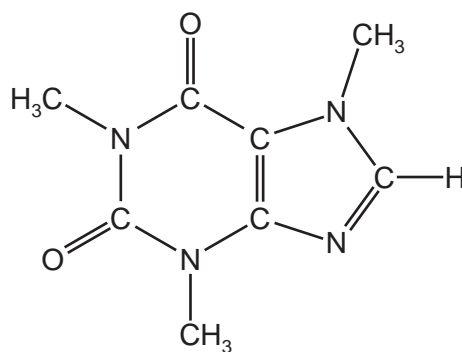


8 In which lists are the compounds in the same homologous series?

- 1 CH_4 , C_2H_4 , C_3H_8
- 2 CH_3OH , $\text{C}_2\text{H}_5\text{OH}$, $\text{C}_3\text{H}_7\text{OH}$
- 3 $\text{CH}_3\text{CO}_2\text{H}$, $\text{CH}_3\text{CH}_2\text{OH}$, $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$

- A** 1 and 2 **B** 1 and 3 **C** 2 only **D** 3 only

9 Caffeine is a stimulant found in coffee.



caffeine

Which formula represents caffeine?

- A** $\text{C}_7\text{H}_{10}\text{N}_4\text{O}_2$ **B** $\text{C}_8\text{H}_{10}\text{N}_3\text{O}_2$ **C** $\text{C}_8\text{H}_{10}\text{N}_4\text{O}_2$ **D** $\text{C}_8\text{H}_{11}\text{N}_4\text{O}_2$

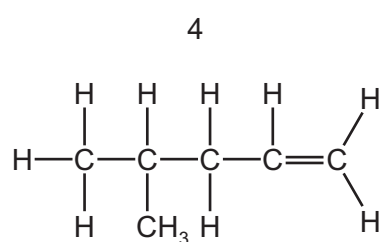
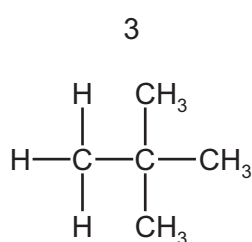
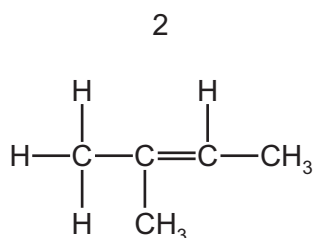
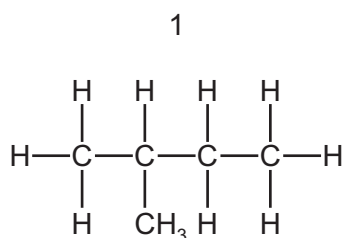
10 Which compound is a member of the alkene homologous series?

- A** C_2H_6 **B** C_4H_{10} **C** C_6H_{12} **D** C_8H_{18}

Paper 2

**Questions are applicable for both core and extended candidates
unless indicated in the question**

11 Which molecules are structural isomers? **(extended only)**



- A** 1 and 2 **B** 1 and 3 **C** 2 and 4 **D** 3 and 4

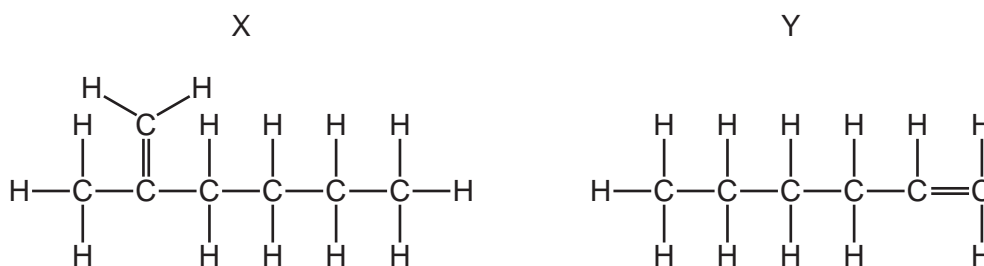
12 The structural formula of compound Q is given.



What is compound Q? **(extended only)**

- A** butyl butanoate
B butyl propanoate
C propyl butanoate
D propyl propanoate

13 The structures of two molecules, X and Y, are shown.



Which row describes X and Y? **(extended only)**

	structural isomers	belong to same homologous series
A	no	no
B	no	yes
C	yes	no
D	yes	yes

14 Which statement defines structural isomers? **(extended only)**

- A** They are compounds with the same displayed formula but a different molecular formula.
- B** They are compounds with the same molecular and displayed formulae but a different structural formula.
- C** They are compounds with the same molecular formula but a different structural formula.
- D** They are compounds with the same structural formula but a different displayed formula.

15 Which statement about structural isomers is correct? **(extended only)**

- A** They have the same structure but different reactivity.
- B** They have the same general formula but a different number of carbon atoms in their molecules.
- C** They have the same structure but different relative molecular masses.
- D** They have different structures but the same numbers of each type of atom.

16 Which formula is the same in methanol, ethanol and propanol?

- A** empirical formula
- B** general formula
- C** molecular formula
- D** structural formula